		Course Code	: MP	H 220	1
		O P JINDAL UNIVERSITY		S. Tal	OPE
	ig tall	Mid Semester Examination, April-2023			OPJU
		M.Sc. (Physics) 4 th Semester			resa Telaponerio Cultura
		DEPARTMENT OF PHYSICS			
		NUCLEAR AND PARTICLE PHYSICS [03PG011]			
NT-4-		e: 2 Hrs. Ma	x. Ma	rks: 50)
Note			NA	CO	IZI
		C 4 4 (20 1)	M	CO	KI
		Section A (20 marks)			
	T	Answer any 4 questions [04 x 05 marks = 20 marks]			
	a.	Describe Yukawa theory of nuclear forces. Discuss how the interaction energy between p-p or p-n is calculated.	4	CO1	K2
	b.	What is threshold energy? Write down the formula of threshold energy. At			
		what angle will threshold energy be maximum?	4	CO2	K1
1	c.	Discuss why nuclear binding energy is higher than atomic binding energy.	4	CO1	K2
	d.	What does the X in ${}^{14}_{7}N(X,p){}^{17}_{8}O$ represent? Derive an expression of nuclear reaction cross section.	4	CO2	K2
	e.	Discuss that nuclear forces are saturated and non-central. What do you mean by isospin and parity of nucleus.	4	CO1	K2
		Section B (30 marks)			
		Answer any 3 questions [03 x 10 marks=30 marks]			
	a.	Explain the terms: mass defect and binding energy of the nucleaus. Give the			
		methods of calculation of binding energy for a nucleus. Also, give main	10	CO1	K3
		features of binding energy per nucleon versus mass number graph.	10		110
	b.	Consider the bound state deuterium nucleus. Determine its ground state energy using quantum mechanical model.	10	CO1	K3
2	c.	Prove that the threshold energy of a nuclear reaction $(E_1)_{th} = (M_1 + M_2)/M_2$ where, M_1 is mass of incident particle and M_2 is mass of the target nucleus.	10	CO2	K3
_	d.	If ${}^{3}H$ is bombarded with 2.5 MeV protons, what are the energies of the neutrons produced at 60°. Given mass of ${}^{1}H = 1.00782503$ amu, mass of $n = 1.008664915$ amu, mass of ${}^{3}H = 3.01602932$ amu.	10	CO2	К3
	e.	Obtain the threshold energy for reaction $^{27}_{13}Al(\alpha, p)^{31}_{15}P$ Given: Mass of $^{27}_{13}Al = 25.986892$ amu, Mass of $\alpha = 4.002603$ amu, mass of p = 1.007825 amu, mass of $^{31}_{15}P = 30.973765$ amu	10	CO2	K3

Course Code: 03PG011 O P JINDAL UNIVERSITY **OPJU** Mid Semester Examination, April-2023 M.Sc. 4th Semester **PHYSICS** COMMUNICATION ELECTRONICS-II [MPH2204] Time: 2 Hrs. Max. Marks: 50 Note: M CO | KL Unit-I (20 marks) Answer any 4 questions [05 x 04 marks=20 marks] What do you mean by pulse modulation? Also write its four advantages. 4 1 1 Explain pulse position modulation. Focus on its advantage and disadvantage b. 4 1 2 over other modulation system. What is the method, to calculate sampling frequency of a band pass signal? 1 4 1 2 Explain with giving appropriate example. What do you understand by Delta modulation. d. 4 1 e. Explain Deferential pulse code modulation (DPCM). 4 1 Unit-II (30 marks) Answer any 3 questions [03 x 10 marks=30 marks] How a continues time signal convert to a discrete signal by using sampling 10 1 3 technique. What do you understand by pulse code modulation? Give its advantage and b. 10 1 3 disadvantage. 2 What is natural sampling explain. c. 10 1 3

2

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What do you mean by Phase shift keying (PSK).

What do you mean by Phase shift keying (FSK).

d.

e.

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		O P JINDAL UNIVERSITY Mid-Semester Examination, April-2023			OPJU
		M.Sc. 4 th Semester			o Tronvesoro
		LASER PHYSICS AND APPLICATIONS		SSO MANAC	2006083
		[SOS-M-PH2205]			
	Time		Jax. N	Aarks: 50)
Not	e:				
			M	СО	K L
		Section A (10 marks) All Questions are compulsory [05 x 02 marks=10 marks]			
	a.	Name some properties, which make laser light different from ordinary light.	02	CO-1	K1
	b.	What is population inversion? How is it achieved?	02	CO-1	K1
1	c.	Name three methods of pumping a Laser.	02	CO-1	K1
	d.	Explain the term Spatial and Temporal coherence.	02	CO-1	K
	e.	The output power of a given laser is 1mW and the emitted wavelength is 630nm. Calculate the number of photons emitted per second. If the area of the laser beam is 10 ⁻⁶ m ² , then find intensity of laser beam.	02	CO-1	K2
		Section B (16 marks) Answer any 4 questions [04 x 04 marks=16 marks]			
	a.	Derive the relationship between various Einstein's coefficients. What are the necessary conditions for the laser action to take place?	04	CO-1	K2
	b.	Define the Laser. Also, explain the Principles and conditions of laser action.	04	CO-1	K1
2	c.	A laser beam can be focused on an area equal to the square of its wavelength λ^2 . For a He-Ne laser λ = 6328 A^0 . If the laser radiates energy at the rate of 1mW, find out the intensity of the focused beam.	04	CO-1	K1
	d.	What is resonator? Explain the role of feedback in a laser cavity.	04	CO-2	K1
	e.	Explain the working, construction, and energy level diagram for Semiconductor Laser.	04	CO-3	K1
		Section C (24 marks) Answer any 3 questions [03 x 08 marks=24 marks]			
	a.	Explain the Construction and Operation of Solid State Laser.	08	CO-3	K1
	b.	What is the Pumping process? Also define three- level, and four- level laser pumping.	08	CO-1	K1
	c.	Briefly discuss the construction and work in helium-neon laser with the energy level diagram.	08	CO-3	K1
3	d.	What do you mean by the terms stimulated absorption, spontaneous emission, and stimulated emission.	08	CO-1	K1
	e.	(i) Find the momentum and energy of a photon of 6328A ⁰ wavelength of laser beam.		CO-1	K2
		(ii) If the wavelength of sodium line 5890A ⁰ and coherent length 2.5cm. Calculate coherent time, half width of spectral line and quality of spectral line	08		

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O P JINDAL UNIVERSITY
Mid Semester Examination, April-2023
M.Sc. 4th Semester
CHEMISTRY

OPJU

Course Code: MCH 2201

	Tim	ENVIRONMENTAL CHEMISTRY [03PG012] e: 2 Hrs.	- M-	-1 50	
Note		mpt both sections	X. Ma	rks: 50	
-111 milit			M	CO	KI
		Section A (20 marks) Answer any 4 questions [05 x 04 marks=20 marks]			
	a.	Define Atmosphere. Discuss important roles and services provided by it.	5	1	2
	b.	Write a note on Earth's Energy Balance.	5	1	2
1	c.	Define Hydrosphere. Compare the composition & properties of ground and surface water.			
	d.	Classify water pollutants and briefly discuss their ill effects.	5	2	2
	e.	Differentiate between BOD and COD, as used for accessing the quality of water.	5	2	3
		Section B (30 marks) Answer any 3 questions [03 x 10 marks=30 marks]			
	a.	Draw the Temperature profile of the Atmosphere. Discuss the composition and salient feature of its segments.	10	1	2
*	b.	What is temperature Inversion? Discuss the chemical and photochemical reactions that results in the formation of Photochemical Products. Mention some of their harmful effects.	10	3	3
2	c.	Discuss the various physico-chemical processes involved during the cycling of the following elements between various parts of the Earth. a. Nitrogen b. Sulphur	10	3	3
	d.	Discuss how Dissolved oxygen is used as an important parameter for analyzing the quality of water. Explain Winkler method for its determination.	10	2	3
	e.	Briefly outline the various stages of waste water treatment. Discuss the steps involved during the treatment of wastewater by Activated Sludge process and write its merits and demerits.	10	4	3

1		Course Code:	MCI	12203	
		O P JINDAL UNIVERSITY			OPIU
		Mid Semester Examination, April-2023			
		M.Sc. 4th Semester		Occupant of the Market	Balli Branasaka Balingan
<u> </u>		M.Sc. Chemistry			
	Tim	Heterocyclic Chemistry [03PG012]			
	11111	e: 2 Hrs. Ma		rks: 50	1
		Section A (10 marks)	M	СО	KL
	a.	All Questions are compulsory [05 x 02 marks=10 marks] Write the name of given heterocyclic compounds on the basis of replacement nomenclature:	2	1	1
1	b.	Draw the structure of following heterocyclic compounds: 1. Oxetane 2. Theipane	2	1	1
	c.	Write the name of given heterocyclic compounds on the basis of Hantzsch-Widman nomenclature:	2	1	1
	d.	Write the criteria of aromaticity for a heterocyclic compound.	2	1	1
	e.	Explain delocalization energy.	2	2	1
		Section B (16 marks) Answer any 4 questions [04 x 04 marks=16 marks]			
	a.	What do you understand by ring inversion?	4	2	2
	b.	Classify the heterocyclic compounds on the basis of their structure with examples.	4	1	2
2	c.	Explain the 1,3-Di-axial interaction in six membered heterocyclic compounds.	4	2	2
	d.	What do you understand by torsional strain.	4	2	2
	e.	Write the general chemical behaviour of heterocyclic compounds.	4	1	1
		Section C (24 marks) Answer any 3 questions [03 x 08 marks=24 marks]		•	•
	a.	Write note on: 1. Empirical Resonance Energy 2. Dewar Resonance Energy	8	2	2
	b.	Explain the heteroaromatic reactivity and tautomerism in aromatic heterocycles.	8	2	3
3	c.	Explain the principles of heterocyclic synthesis involving cyclization reactions and cycloaddition reactions.	8	2	3
	d.	Explain the systematic nomenclature (Hantzsch-Widman system) for monocyclic heterocycles.	8	1	2
	e.	Draw and explain the conformation of six-membered heterocycles with reference to molecular geometry and pyramidal inversion.	8	2	3

		Course Code	e: MC	CH222	2
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		Mid Semester Examination, April-2023			OPJU
	11/11/	MSc 4th Semester			den Turksbook Narmen
		MSc .Chemistry			
		Chemistry of Natural Products [03PG012]			
		e: 2 Hrs. Ma	x. Ma	rks: 50	
Note	•	1 (1988) 기업 경기 (1989) 전 1 (1988) 1 (1989) 1 (1989) 1 (1989) 기업 (1989) 1 (1989) 1 (1989) 1 (1989) 1 (1989) 1 (1 1989) 1 (1989) 1 (1989) 1 (1989) 1 (1989) 1 (1989) 1 (1989) 1 (1989) 1 (1989) 1 (1989) 1 (1989) 1 (1989) 1 (1		T ~~	
			M	CO	KL
		Section A (20 marks) Answer any 4 questions [05 x 04 marks=20 marks]			
	a.	What do you mean by isoprene rule for terpenes? Explain.	4	1	1
	b.	Elucidate structure of menthol	4	4	2
	c.	How will you convert: 3 di-bromo -3-methyl butane into citral?	4	2	2
1	d.	How will you convert.			
		(i) Limonene to corvoxime. (ii) L-terpineol to limonene.	4	2	2
	e.	Define carotenes? Write its method of structure determination	4	4	2
		Section B (30 marks)			
	T	Answer any 3 questions [03 x 10 marks=30 marks]			
	a.	What are terpenoids? How are they classified? How are these extracted? Write a short note on their chemical nature.	10	1	1
2	b.	Discuss the constitution, properties and uses of menthol as limonene	10	3	3
2	c.	Give general method of structure elucidation of terpenoids?	10	4	1
	d.	What class of carotenes are termed as Xanthphylls? Explain any two xanthophylls.	10	1	1

Course Code: MMA2201

O P JINDAL UNIVERSITY

Mid Semester Examination, April-2023 MSc 4th Semester



MSc (Math)

Artificial	Intelligence a	nd Neural	Network	IMMA22011
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NT .		e: 2 Hrs. Ma	x. Ma	rks: 50	
Note			M	СО	KI
		Section A (10 marks) All Questions are compulsory [05 x 02 marks=10 marks]	eng verter statementer	A Marco advances a river y transfer to	
	a.	Define artificial intelligence	2	1	1
	b.	Explain the concept of State Space search.	2	1	1
1	c.	Define a Production system in artificial intelligence.	2	1	1
	d.	Explain the significance of knowledge representation in artificial intelligence.	2	2	1
	e.	Outline the levels of representation in knowledge representation.	2	2	1
*		Section B (16 marks) Answer any 4 questions [04 x 04 marks=16 marks]		1	
11	a.	Define a Game Tree and its significance in artificial intelligence.	4	1	2
1	b.	Identify different types of problem spaces in artificial intelligence.	4	1	2
	c.	Define Hill climbing algorithm in artificial intelligence	4	1	2
2	d.	What is knowledge representation in artificial intelligence, and what are its different levels of representation?	4	2	2
	e.	Can you explain Formal logic, and how is it used in knowledge representation in artificial intelligence?	4	2	2
		Section C (24 marks) Answer any 3 questions [03 x 08 marks=24 marks]	t many emi-partition repetition (A CONTROL OF THE STATE OF THE S	ana tikuwinga alifunga ku akin
	a.	Discuss the Breadth first Search technique in artificial intelligence and provide examples of its potential applications.	8	1	3
	b.	Explain how Best first functions as a heuristic search technique in artificial intelligence problem-solving and discuss its strengths and limitations.	8	1	3
3	c.	Describe the functioning of the Min Max algorithm in artificial intelligence, along with its pros and cons.	8	1	3
	d.	How does an Inference Engine work in artificial intelligence, and what are some of its key features and capabilities?	8	2	3
	e.	How do knowledge representation schemes, such as Semantic nets, Frames, and Scripts, function in artificial intelligence, and what are some examples of their use?	8	2	3

in h								Course (Code:	MMA	2200
						Charles A 10 Ch	VERSIT				OPIL
		1	Mid S			aminati	ion, April-2	2023	To be starting	Unnison or	STEEL TRESSURE
						nematic				AND ME	MARKEN
			O	perat			[03PG013	1	gard.		
im	ie: 2 H	Irs.							Max	. Marl	ks: 50
									M	CO	KI
			Aı	nswer			20 mark [04 x 05 ma	rks=20 marks]			
	a.	restrictions, it c B in a week. The to produce a to	annot pere are on of per	oroduc 160 pr roduct ton. Fo	e more oductions ormulat	than 40 on hours d B resp	0 tons of gra in a week. It pectively wit	B. Because of raw material ade A and 300 tons of grade t requires 0.2 and 0.4 hours th corresponding profits of to maximize the profit and	5	1	2
	b.	Solve using gr	aphical	meth	od:						
		Maximize z =	$3x_1 + 4$	x_2							
		$subject to 5x_1 + 4x_2 \le 200$									
	-	$3x_1$	$3x_1 + 5x_2 \le 150$						5	1	2
		$5x_1$	$+4x_{2}$	≥100							_
			$+4x_{2}$								
			$c_2 \ge 0$								
	c.	Solve using sin		netho	d:						
		Minimize z =									
		subject to x_1									
			$+2x_{2}$						5	1	2
			$+8x_{2}$								
			$x_2 \ge 0.$								
	d.			sible	Solutio	on of th	e following	Transportation problem			
		using Matrix M				on or th	o lollowing	, Transportation problem			
		Ū	D ₁	D_2	D_3	D ₄	Supply				
		Oı	21	16	25	13	11		5	1	2
		O ₂	17	18	14	23	13				_
		O ₃	32	17	18	41	19				
		Demand	6	10	12	15					
The	e.	Write notes on	invento	orv co	st				5	6	1

7 (18) 7 (18)			A	Answer	Un	nit-II (30 questions [03]	marks) 10 marks=30 marks]		a de la composición della comp	
	a.	Solve the LP	P using	Big N	A meth	nod:	TO marks 30 marks	Samon Sa Samon Samon Sa	1	
		Minimize z				9000		anaria di dalam		
ļ.		subject to								
	i Agiya		$4x_1 + 3$					10	1	2
			$x_1 + 2x$							
			$x_1, x_2 \geq$							
	b.	Using dual sin	•		d to so	lve:				
		Minimize z =	$= 5x_1 +$	$6x_{2}$	a 10 30					
	langer.	$subject to x_1$								
			$x_1 + x_2 = x_1 + x_2$					10	2	2
			$x_1, x_2 \ge$					and the state of		
_	C	c. Apply the principle of duality to solve:								
2	"									
		$\begin{aligned} & \textit{Maximize } z = 6x_1 + 8x_2 \\ & \textit{subject to } 5x_1 + 2x_2 \le 20 \end{aligned}$								
		한 사용병하는 경상이 불레스 보다 먹었다						10	2	2
			$+2x_2$					- jet i sa se pasisis	M.	
	-		$x_1, x_2 \ge$							
	d.	Solve the Tran		-						
		O ₁	D_1	D ₂	D ₃	Supply				
		O_1	2	4	7	15		4 - 58		
		O_2	3 5	3	1	18		10	3	2
		O ₃	$\frac{3}{1}$	6	7	17				
		Demand	17	19	28	14				
	e.					-'41				
		Write notes on	Purcha	sing n	nodel v	with no shorta	iges.	10	6	2

CALL CALL A			Answe	er anv 3	nit-II (30 marks) questions [03 x 10 marks=30 r	1.1		
a.		1 1 usii	gra gr	M met	hod:	narksj		
	Minimize	z = 4x	$_{1}+x_{2}^{-}$					
	subject to			3				
1 194		$4x_1 +$				10	1	
		$x_1 + 2$	24 15 17 17 17					
		$x_1, x_2 \ge$						
b.	Using dual s			od to so	ilve:			
	Minimize z	$=5x_1 +$	6x	/G 10 50				
	subject to x							
		$x_1 + x_2$				10	2	
		$x_1, x_2 \ge$					30-1	
C.	Apply the pri							
	Maximize z							
	subject to 53							
	x_{l}	10	2					
		$x_1, x_2 \geq$						
d.	Solve the Tran	The state of the s		oblem			W. Co	
13		Di	D ₂	D ₃	Supply			
	O_1	2	4	7	15			
	O ₂	3	3	1	18			
	O ₃	- 5	4	7	17	10	3	2
	O ₄	1	6	2	14			
100 100 100 100 100 100 100 100 100 100	Demand	17	19	28				
e. V		V. 19 11 11 11 11 11 11 11 11 11 11 11 11			rith no shortages.		9 1 1 1	

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		Mid Semester Examination, April-2023			Orjo
		M.Sc. 4 th Semester		University of Se (SOMA)	HI HONGES
		Department of Mathematics			
	Ti-	Cryptography and Network Security [03PG013]		1	
	1 111	ne: 2 Hrs.	ax. M	arks: 50	
			M	CO	KL
		Section A (20 marks)			122
		Answer any 4 questions [05 x 04 marks=20 marks]			
7					
.}	a.	Encrypt the message "RETURN HOME" using the Caesar Cipher.	5	CO2	K2
	b.	Explain Diffie-Hellman Key Exchange Cryptosystem.	5	CO5	K3
1	c.	Encrypt The Message "CRYPTO" using key pair (5, 3) in Affine Cipher.	5	CO2	K2
7	d.	Explain types of attack in public key cryptography.	5	CO3	K2
2	e.	Explain RSA Algorithm.	5	CO4	K3
		Section B (30 marks)			
		Answer any 3 questions [03 x 10 marks=30 marks]			
	a.	Explain ElGamal Signature Scheme with Model.	10	CO5	К3
2	b.	Perform encryption and decryption using the RSA Algorithm, if p=3, q=13, e=5 and M=10.	10	CO4	K2
	c.	Explain Asymmetric key cryptography and Symmetric key cryptography.	10	CO3	K2
	d.	Define Digital Signature scheme and also explain RSA Signature Scheme.	10	CO5	K3

		Course Co	de: Mi	MA220	2
4.20		O. P. JINDAL UNIVERSITY		1	
		Mid Semester Examination, April-2023			OPIL
		M.Sc. 4 th Semester		Uniterate of	STEEL TECHNOLOGY
7,75		SCHOOL OF SCIENCE			
F1-17	Tim	DATABASE MANAGEMENT SYSTEM[03PG013] ne: 2 Hrs.	to Appendicus		
Note		N	Iax. Ma	rks: 50)
			M	CO	KL
		Section A (20 marks) Answer any 4 questions [05 x 04 marks=20 marks]		100	
	a.	What is Database Management System? How it is different from file System?	5	1	1
	b.	Classify the Components of Relational Data Model.	5	2	1
1	c.	What is Generalization, Specialization & Aggregation?	5	1	-
	d.	What is Schema & Instances?		1	1
	e.	Differentiate between week & strong entity?	5	1	1
	1	Section B (30 marks) Answer any 3 questions [03 x 10 marks=30 marks]]]	1	
	a.	Explain ER Models with Entity & Attributes.	10	1	2
2	b.	Explain Integrity Constraint.	10	2	77.11
	c.	Construct relational table by ER-Diagram.	+		2
	d.	Discuss DDL & DML with Example Query.	10	1	2

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